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Aqua[*N*-phenyl-2-(quinolin-8-yloxy)-acetamide]dinitratozinc(II)

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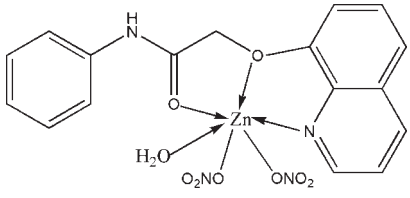
Key indicators: single-crystal X-ray study; T = 296 K; mean σ(C–C) = 0.004 Å; R factor = 0.034; wR factor = 0.093; data-to-parameter ratio = 14.3.

In the title complex, $[Zn(NO_3)_2(C_{17}H_{14}N_2O_2)(H_2O)]$, the six-coordinated Zn atom is in a distorted octahedral geometry, the donor centers being two O atoms and one N atom from the tridentate organic ligand, a water O atom and two O atoms from two monodentate nitrate ions. In the crystal, O—H···O hydrogen bonds between the coordinated water molecules and nitrate O atoms and N—H···O hydrogen bonds between the main ligand and nitrate O atoms consolidate the three-dimensional network.

Related literature

For the synthesis of *N*-phenyl-2-(quinolin-8-yloxy)acetamide, see: Li *et al.* (2005); Wu *et al.* (2006). For the crystal structure of the hydrate of this molecule, see: Li *et al.* (2005). For the coordination ability of related amides to lanthanides, see: Cai & Tan (2002); Wu *et al.* (2006).



Experimental

Crystal data

$[Zn(NO_3)_2(C_{17}H_{14}N_2O_2)(H_2O)]$
M_r = 485.71
Triclinic, $P\overline{1}$
a = 7.9980 (14) Å
b = 9.5109 (16) Å
c = 13.359 (2) Å
α = 94.876 (2)°
β = 96.496 (2)°
γ = 106.031 (2)°
V = 963.2 (3) Å^3
Z = 2
Mo $K\alpha$ radiation
μ = 1.34 mm^{-1}
T = 296 K
0.30 × 0.19 × 0.11 mm

Data collection

Bruker SMART CCD diffractometer
Absorption correction: multi-scan (*SADABS*; Bruker, 1997)
T_{min} = 0.745, T_{max} = 0.863
10541 measured reflections
4030 independent reflections
3454 reflections with $I > 2\sigma(I)$
R_{int} = 0.028

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.034
$wR(F^2)$ = 0.093
S = 1.05
4030 reflections
281 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.66 e Å^{-3}
$\Delta\rho_{min}$ = −0.39 e Å^{-3}

Table 1
Selected geometric parameters (Å, °).

Zn1—O6	2.0231 (17)	Zn1—O2	2.1110 (15)
Zn1—O3	2.0494 (17)	Zn1—N1	2.1166 (17)
Zn1—O9	2.0818 (17)	Zn1—O1	2.2612 (16)
O6—Zn1—O3	103.05 (7)	O9—Zn1—N1	87.29 (7)
O6—Zn1—O9	86.57 (8)	O2—Zn1—N1	145.38 (7)
O3—Zn1—O9	170.28 (7)	O6—Zn1—O1	163.42 (7)
O6—Zn1—O2	93.07 (7)	O3—Zn1—O1	86.10 (7)
O3—Zn1—O2	95.75 (7)	O9—Zn1—O1	84.93 (7)
O9—Zn1—O2	85.04 (7)	O2—Zn1—O1	72.04 (6)
O6—Zn1—N1	120.13 (8)	N1—Zn1—O1	73.67 (6)
O3—Zn1—N1	86.60 (7)		

Table 2
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
O9—H9A···O6^{i}	0.82	1.99	2.803 (3)	173
O9—H9B···O4^{ii}	0.88	1.97	2.797 (3)	155
N2—H2···O8^{iii}	0.86	2.07	2.869 (3)	155

Symmetry codes: (i) $-x+1, -y+2, -z$; (ii) $x+1, y, z$; (iii) $x, y-1, z$.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BH2270).

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supplementary materials

Acta Cryst. (2010). E**66**, m288 [**doi:10.1107/S160053681000471X**]

Aqua[*N*-phenyl-2-(quinolin-8-yloxy)acetamide]dinitratozinc(II)

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Comment

Including our previous work (Wu *et al.*, 2006), the amide type ligands have been widely used to enhance the luminescent emissions of the lanthanide ions because of their excellent coordination ability (Cai & Tan, 2002). However, little work has been done on their transition metal complexes. Therefore, as part of our ongoing studies of the amide type ligands, the title complex was synthesized and characterized by X-ray diffraction.

As shown in Fig. 1, in the title complex, the six-coordinated Zn atom is in a distorted octahedral geometry with the donor centers of two O atoms and one N atom from the ligand, one O atom from one water molecule and two O atoms from two nitrate ions. The dihedral angle between phenyl ring and naphthyl ring is 22.5 (1)°.

In the crystal, O—H···O hydrogen bonds between the coordinated water molecules and nitrate O atoms, and N—H···O hydrogen bonds between the ligands and nitrate O atoms are helpful to consolidate the three-dimensional network (Fig. 2).

Experimental

N-phenyl-2-(quinolin-8-yloxy)acetamide (Wu *et al.*, 2006; Li *et al.*, 2005) (0.278 g, 1 mmol) was dissolved in acetonitrile (10 ml), then an acetonitrile solution (10 ml) containing zinc nitrate hexahydrate (0.295 g, 1 mmol) was added dropwise at room temperature. After stirring for 2 h, the mixture was filtered and set aside to crystallize at room temperature for 8 d, giving colorless prismatic crystals.

Refinement

Atoms H9A and H9B (water molecule O9) were located in a difference map and their positions fixed, with U_{iso}(H) = 1.5U_{eq}(O9). Other H atoms bonded to C and N atoms were placed in calculated positions and treated using a riding-model approximation [aromatic groups: C—H = 0.93 Å and U_{iso}(H) = 1.2U_{eq}(carrier C); methylene group: C—H = 0.97 Å and U_{iso}(H) = 1.2U_{eq}(carrier C); amine group: N—H = 0.86 Å and U_{iso}(H2) = 1.2U_{eq}(N2)].

Figures

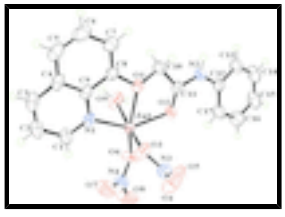

Fig. 1. The molecular structure shown with 50% probability displacement ellipsoids.

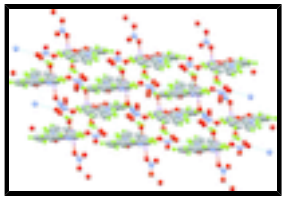

Fig. 2. The crystal packing for the title complex *via* hydrogen bonds (dashed lines).

Aqua[N-phenyl-2-(quinolin-8-yloxy)acetamide]dinitratozinc(II)

Crystal data

$[Zn(NO_3)_2(C_{17}H_{14}N_2O_2)(H_2O)]$	$Z = 2$
$M_r = 485.71$	$F(000) = 496$
Triclinic, $P\overline{1}$	$D_x = 1.675$ Mg m^{-3}
Hall symbol: -P 1	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
$a = 7.9980$ (14) Å	Cell parameters from 4639 reflections
$b = 9.5109$ (16) Å	$\theta = 2.6$–$26.4°$
$c = 13.359$ (2) Å	$\mu = 1.34$ mm^{-1}
$\alpha = 94.876$ (2)°	$T = 296$ K
$\beta = 96.496$ (2)°	Prism, colorless
$\gamma = 106.031$ (2)°	0.30 × 0.19 × 0.11 mm
$V = 963.2$ (3) Å^3	

Data collection

Bruker SMART CCD diffractometer	4030 independent reflections
Radiation source: fine-focus sealed tube	3454 reflections with $I > 2\sigma(I)$
graphite	$R_{int} = 0.028$
φ and ω scans	$\theta_{max} = 26.7°$, $\theta_{min} = 2.2°$
Absorption correction: multi-scan (SADABS; Bruker, 1997)	$h = -10 \rightarrow 10$
$T_{min} = 0.745$, $T_{max} = 0.863$	$k = -11 \rightarrow 11$
10541 measured reflections	$l = -16 \rightarrow 16$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.034$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.093$	H-atom parameters constrained
$S = 1.05$	$w = 1/[\sigma^2(F_o^2) + (0.0544P)^2 + 0.1851P]$ where $P = (F_o^2 + 2F_c^2)/3$
4030 reflections	$(\Delta/\sigma)_{max} < 0.001$
281 parameters	$\Delta\rho_{max} = 0.66$ e Å^{-3}

0 restraints $\Delta\rho_{min} = -0.39$ e Å^{-3}

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
Zn1	0.40610 (3)	0.96202 (3)	0.180722 (18)	0.03178 (10)
O1	0.3965 (2)	0.74601 (17)	0.24213 (12)	0.0397 (4)
O2	0.2694 (2)	0.78947 (17)	0.06551 (12)	0.0379 (4)
O3	0.1934 (2)	0.9611 (2)	0.25329 (13)	0.0468 (4)
O4	−0.0719 (3)	0.9596 (3)	0.26368 (17)	0.0721 (6)
O5	0.0289 (3)	0.9897 (3)	0.12235 (16)	0.0820 (8)
O6	0.4097 (3)	1.1192 (2)	0.08737 (14)	0.0524 (5)
O7	0.3915 (4)	1.2818 (2)	0.20478 (17)	0.0757 (7)
O8	0.3628 (4)	1.3175 (2)	0.04674 (17)	0.0769 (7)
O9	0.6309 (2)	0.9394 (2)	0.12457 (14)	0.0566 (5)
H9A	0.6102	0.9205	0.0626	0.085*
H9B	0.7372	0.9403	0.1513	0.085*
N1	0.5583 (2)	1.0166 (2)	0.32636 (13)	0.0337 (4)
N2	0.2226 (3)	0.5535 (2)	−0.00405 (15)	0.0401 (5)
H2	0.2306	0.4695	0.0117	0.048*
N3	0.0477 (3)	0.9699 (2)	0.21100 (16)	0.0447 (5)
N4	0.3863 (3)	1.2425 (2)	0.11447 (17)	0.0480 (5)
C1	0.6333 (3)	1.1531 (3)	0.36958 (19)	0.0450 (6)
H1	0.6103	1.2301	0.3375	0.054*
C2	0.7469 (4)	1.1865 (3)	0.4625 (2)	0.0530 (7)
H2A	0.7961	1.2840	0.4912	0.064*
C3	0.7839 (4)	1.0767 (3)	0.50957 (19)	0.0528 (7)
H3	0.8605	1.0985	0.5703	0.063*
C4	0.7068 (3)	0.9290 (3)	0.46704 (17)	0.0431 (6)
C5	0.7382 (4)	0.8075 (4)	0.5099 (2)	0.0566 (7)
H5	0.8126	0.8223	0.5711	0.068*
C6	0.6612 (4)	0.6687 (4)	0.4632 (2)	0.0608 (8)
H6	0.6842	0.5897	0.4927	0.073*
C7	0.5464 (4)	0.6413 (3)	0.37011 (19)	0.0510 (7)
H7	0.4958	0.5456	0.3383	0.061*
C8	0.5119 (3)	0.7573 (3)	0.32834 (17)	0.0370 (5)
C9	0.5925 (3)	0.9046 (3)	0.37403 (16)	0.0343 (5)
C10	0.3655 (3)	0.6234 (3)	0.16737 (18)	0.0411 (5)
H10A	0.2877	0.5363	0.1880	0.049*
H10B	0.4750	0.6033	0.1567	0.049*
C11	0.2808 (3)	0.6650 (2)	0.07116 (17)	0.0339 (5)
C12	0.1501 (3)	0.5535 (3)	−0.10558 (18)	0.0381 (5)
C13	0.1263 (4)	0.4259 (3)	−0.1706 (2)	0.0494 (6)
H13	0.1577	0.3465	−0.1465	0.059*
C14	0.0558 (4)	0.4166 (3)	−0.2714 (2)	0.0610 (8)
H14	0.0408	0.3311	−0.3150	0.073*
C15	0.0081 (4)	0.5329 (3)	−0.3068 (2)	0.0629 (8)
H15	−0.0394	0.5267	−0.3744	0.076*

C16	0.0308 (4)	0.6595 (3)	−0.2417 (2)	0.0604 (8)
H16	−0.0018	0.7382	−0.2661	0.072*
C17	0.1011 (4)	0.6711 (3)	−0.1408 (2)	0.0493 (6)
H17	0.1154	0.7566	−0.0974	0.059*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Zn1	0.03517 (16)	0.03244 (16)	0.02720 (15)	0.01224 (11)	−0.00243 (10)	0.00149 (10)
O1	0.0471 (9)	0.0359 (9)	0.0332 (8)	0.0153 (7)	−0.0107 (7)	−0.0015 (7)
O2	0.0448 (9)	0.0303 (8)	0.0348 (8)	0.0113 (7)	−0.0075 (7)	−0.0007 (6)
O3	0.0357 (9)	0.0719 (12)	0.0367 (9)	0.0229 (8)	0.0013 (7)	0.0082 (8)
O4	0.0428 (11)	0.1124 (19)	0.0713 (15)	0.0334 (12)	0.0168 (10)	0.0182 (13)
O5	0.0611 (14)	0.148 (2)	0.0473 (13)	0.0476 (15)	−0.0026 (10)	0.0271 (14)
O6	0.0806 (13)	0.0421 (10)	0.0500 (11)	0.0342 (10)	0.0243 (10)	0.0164 (8)
O7	0.115 (2)	0.0597 (14)	0.0523 (13)	0.0335 (13)	0.0008 (12)	−0.0042 (11)
O8	0.126 (2)	0.0575 (13)	0.0664 (14)	0.0540 (14)	0.0119 (13)	0.0232 (11)
O9	0.0378 (10)	0.0931 (15)	0.0413 (10)	0.0280 (10)	0.0000 (8)	−0.0016 (10)
N1	0.0325 (10)	0.0397 (11)	0.0271 (9)	0.0100 (8)	−0.0001 (7)	0.0016 (8)
N2	0.0509 (12)	0.0290 (10)	0.0365 (11)	0.0114 (9)	−0.0060 (9)	−0.0010 (8)
N3	0.0375 (11)	0.0521 (13)	0.0450 (12)	0.0175 (9)	−0.0009 (9)	0.0022 (10)
N4	0.0587 (14)	0.0367 (11)	0.0504 (13)	0.0177 (10)	0.0046 (11)	0.0070 (10)
C1	0.0481 (14)	0.0432 (14)	0.0376 (13)	0.0074 (11)	0.0004 (11)	−0.0016 (11)
C2	0.0539 (16)	0.0532 (16)	0.0370 (14)	−0.0016 (13)	−0.0032 (12)	−0.0062 (12)
C3	0.0450 (15)	0.0700 (19)	0.0307 (13)	0.0029 (13)	−0.0070 (11)	−0.0007 (12)
C4	0.0370 (13)	0.0630 (16)	0.0257 (11)	0.0110 (11)	−0.0010 (9)	0.0032 (11)
C5	0.0592 (17)	0.079 (2)	0.0327 (13)	0.0256 (15)	−0.0087 (12)	0.0133 (14)
C6	0.077 (2)	0.068 (2)	0.0456 (16)	0.0351 (17)	−0.0035 (14)	0.0193 (14)
C7	0.0645 (17)	0.0495 (15)	0.0398 (14)	0.0212 (13)	−0.0035 (12)	0.0092 (12)
C8	0.0382 (12)	0.0444 (13)	0.0282 (11)	0.0142 (10)	−0.0012 (9)	0.0045 (9)
C9	0.0299 (11)	0.0461 (13)	0.0263 (11)	0.0110 (9)	0.0023 (8)	0.0043 (9)
C10	0.0516 (14)	0.0333 (12)	0.0362 (12)	0.0150 (10)	−0.0056 (10)	−0.0008 (10)
C11	0.0323 (11)	0.0328 (12)	0.0340 (12)	0.0080 (9)	−0.0006 (9)	0.0021 (9)
C12	0.0377 (12)	0.0352 (12)	0.0351 (12)	0.0043 (9)	−0.0016 (9)	−0.0008 (9)
C13	0.0568 (16)	0.0402 (14)	0.0447 (14)	0.0107 (12)	−0.0035 (12)	−0.0051 (11)
C14	0.072 (2)	0.0539 (17)	0.0426 (15)	0.0062 (14)	−0.0029 (14)	−0.0165 (13)
C15	0.075 (2)	0.0621 (19)	0.0345 (14)	0.0007 (15)	−0.0108 (13)	−0.0008 (13)
C16	0.073 (2)	0.0503 (17)	0.0489 (16)	0.0117 (14)	−0.0149 (14)	0.0106 (13)
C17	0.0614 (17)	0.0370 (13)	0.0413 (14)	0.0084 (12)	−0.0077 (12)	−0.0008 (11)

Geometric parameters (Å, °)

Zn1—O6	2.0231 (17)	C3—C4	1.408 (4)
Zn1—O3	2.0494 (17)	C3—H3	0.9300
Zn1—O9	2.0818 (17)	C4—C5	1.403 (4)
Zn1—O2	2.1110 (15)	C4—C9	1.418 (3)
Zn1—N1	2.1166 (17)	C5—C6	1.354 (4)
Zn1—O1	2.2612 (16)	C5—H5	0.9300
O1—C8	1.369 (3)	C6—C7	1.418 (4)

O1—C10	1.416 (3)	C6—H6	0.9300
O2—C11	1.220 (3)	C7—C8	1.358 (3)
O3—N3	1.264 (3)	C7—H7	0.9300
O4—N3	1.238 (3)	C8—C9	1.421 (3)
O5—N3	1.214 (3)	C10—C11	1.518 (3)
O6—N4	1.266 (3)	C10—H10A	0.9700
O7—N4	1.225 (3)	C10—H10B	0.9700
O8—N4	1.228 (3)	C12—C17	1.384 (4)
O9—H9A	0.8200	C12—C13	1.385 (3)
O9—H9B	0.8812	C13—C14	1.385 (4)
N1—C1	1.320 (3)	C13—H13	0.9300
N1—C9	1.360 (3)	C14—C15	1.370 (4)
N2—C11	1.339 (3)	C14—H14	0.9300
N2—C12	1.414 (3)	C15—C16	1.379 (4)
N2—H2	0.8600	C15—H15	0.9300
C1—C2	1.409 (4)	C16—C17	1.384 (4)
C1—H1	0.9300	C16—H16	0.9300
C2—C3	1.347 (4)	C17—H17	0.9300
C2—H2A	0.9300		
O6—Zn1—O3	103.05 (7)	C5—C4—C3	124.3 (2)
O6—Zn1—O9	86.57 (8)	C5—C4—C9	119.0 (2)
O3—Zn1—O9	170.28 (7)	C3—C4—C9	116.6 (2)
O6—Zn1—O2	93.07 (7)	C6—C5—C4	120.6 (2)
O3—Zn1—O2	95.75 (7)	C6—C5—H5	119.7
O9—Zn1—O2	85.04 (7)	C4—C5—H5	119.7
O6—Zn1—N1	120.13 (8)	C5—C6—C7	121.5 (3)
O3—Zn1—N1	86.60 (7)	C5—C6—H6	119.3
O9—Zn1—N1	87.29 (7)	C7—C6—H6	119.3
O2—Zn1—N1	145.38 (7)	C8—C7—C6	118.8 (3)
O6—Zn1—O1	163.42 (7)	C8—C7—H7	120.6
O3—Zn1—O1	86.10 (7)	C6—C7—H7	120.6
O9—Zn1—O1	84.93 (7)	C7—C8—O1	124.8 (2)
O2—Zn1—O1	72.04 (6)	C7—C8—C9	121.5 (2)
N1—Zn1—O1	73.67 (6)	O1—C8—C9	113.7 (2)
C8—O1—C10	119.01 (18)	N1—C9—C4	122.7 (2)
C8—O1—Zn1	114.78 (14)	N1—C9—C8	118.74 (19)
C10—O1—Zn1	114.98 (13)	C4—C9—C8	118.6 (2)
C11—O2—Zn1	120.10 (14)	O1—C10—C11	105.98 (18)
N3—O3—Zn1	124.92 (15)	O1—C10—H10A	110.5
N4—O6—Zn1	123.50 (16)	C11—C10—H10A	110.5
Zn1—O9—H9A	109.5	O1—C10—H10B	110.5
Zn1—O9—H9B	135.6	C11—C10—H10B	110.5
H9A—O9—H9B	114.8	H10A—C10—H10B	108.7
C1—N1—C9	118.3 (2)	O2—C11—N2	124.8 (2)
C1—N1—Zn1	123.78 (16)	O2—C11—C10	121.6 (2)
C9—N1—Zn1	117.69 (14)	N2—C11—C10	113.54 (19)
C11—N2—C12	129.5 (2)	C17—C12—C13	120.1 (2)
C11—N2—H2	115.3	C17—C12—N2	123.6 (2)
C12—N2—H2	115.3	C13—C12—N2	116.3 (2)

O5—N3—O4	122.1 (2)	C12—C13—C14	120.1 (3)
O5—N3—O3	120.4 (2)	C12—C13—H13	120.0
O4—N3—O3	117.5 (2)	C14—C13—H13	120.0
O7—N4—O8	123.7 (2)	C15—C14—C13	120.1 (3)
O7—N4—O6	119.6 (2)	C15—C14—H14	120.0
O8—N4—O6	116.7 (2)	C13—C14—H14	120.0
N1—C1—C2	122.5 (3)	C14—C15—C16	119.7 (3)
N1—C1—H1	118.7	C14—C15—H15	120.1
C2—C1—H1	118.7	C16—C15—H15	120.1
C3—C2—C1	119.7 (3)	C15—C16—C17	121.1 (3)
C3—C2—H2A	120.2	C15—C16—H16	119.5
C1—C2—H2A	120.2	C17—C16—H16	119.5
C2—C3—C4	120.1 (2)	C16—C17—C12	119.0 (3)
C2—C3—H3	119.9	C16—C17—H17	120.5
C4—C3—H3	119.9	C12—C17—H17	120.5

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
O9—H9A···O6^{i}	0.82	1.99	2.803 (3)	173.
O9—H9B···O4^{ii}	0.88	1.97	2.797 (3)	155.
N2—H2···O8^{iii}	0.86	2.07	2.869 (3)	155.

Symmetry codes: (i) −*x*+1, −*y*+2, −*z*; (ii) *x*+1, *y*, *z*; (iii) *x*, *y*−1, *z*.

Fig. 1

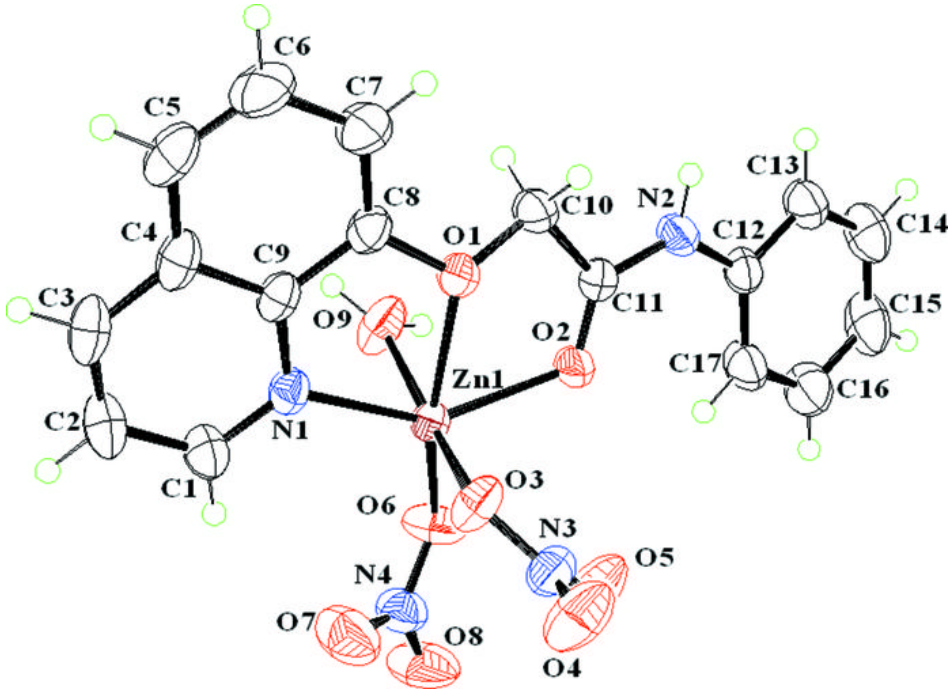
C6
C7
C5
C8
C10
N2
C13
C14
C4
C9
O1
C12
C3
O9
C11
C15
O2
Zn1
C17
C16
C2
N1
C1
O3
O6
N3
O5
N4
O7
O8
O4

Fig. 2